Use of *Tröger*'s Base as a Scaffold for New Chiral Molecular Tweezers: Synthesis of Trimeric, Fused *Tröger*'s Bases¹)

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Dedicated to Professor Rolf Huisgen on the occasion of his 85th birthday

Four novel molecular tweezers, 6-9, have been synthesized having, for the first time, three fused $Tr\ddot{o}ger$'s bases. The compounds differ in the relative configuration of the three fused methylene bridges and have been unambiguously characterized by NMR.

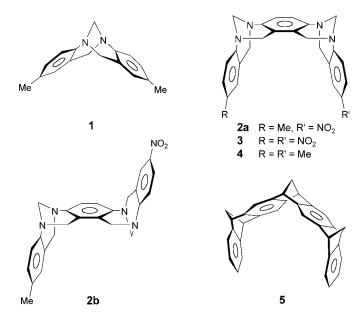
Introduction. – *Tröger*'s base (1) [1] is a concave chiral molecule, the chirality of which results from the blocked configuration of its stereogenic N-atoms. *Tröger*'s base and its derivatives have been described as 'fascinating molecules' [2]. They provide relatively rigid chiral frameworks for the construction of chelating and biomimetic systems, which were essentially developed by *Wilcox* and co-workers, and elaborated by others [3]. *Tröger*'s bases show a perpendicular arrangement of the two aromatic rings [3c], as in *Kagan*'s ether, which was used by *Harmata* and co-workers for the synthesis of molecular tweezers [4]. In this context, we have utilized the *Tröger*'s base skeleton as a scaffold for the construction of the molecular clips 2 [5], 3 [6], and 4 [7].

In this paper, we report the synthesis of chiral molecular tweezers with a trimeric, fused *Tröger*'s base skeleton. During the course of the present work, *Klärner et al.* [8] have described the synthesis of another family of molecular tweezers of type **5**, with three methylene bridges, achiral carbocylic analogues of the compounds described in this paper.

Results and Discussion. – 1. *Synthesis*. The synthesis of the trimeric, fused tweezers **6–8** (*Scheme 1*) started from the dimer **2a**, with relative *syn* configuration of the CH₂ bridges, according to a synthetic pathway similar to the one used for the synthesis of **2a** [5]. Amine **10** was prepared from **2a** by two procedures; in the first one, **10** was obtained in 42% yield by hydrogenation over Pd/C, in the second one, it was prepared in almost quantitative yield by reducing **2a** with SnCl₂ in concentrated HCl. The reaction of **10** with '6-nitroisatoic anhydride'²) in anhydrous THF afforded amide **11** in 66% yield. The latter was reduced to amine **12** (40% yield) by treatment with the

¹⁾ Some preliminary results were described in T. Mas, C. Pardo, J. Elguero, Mendeleev Commun., 2004, 235.

²⁾ Systematic name: 6-nitro-2*H*-3,1-benzoxazine-2,4(1*H*)-dione.



borane-THF complex. Amine **12** was obtained in a higher yield, 71%, by reduction of **11** with the BH₃/Me₂S complex in THF. In the last step, the reaction of **12** with aqueous formaldehyde and concentrated HCl in EtOH at 90° for 40 h yielded a mixture from which were isolated the trimeric bases **6** (11%), **7** (7%), and **8** (8%).

Alternatively, cyclization of **12** with hexamethylenetetramine (HMTA) in trifluoroacetic acid (TFA) at room temperature during 72 h afforded a mixture from which **7** (46%) and **8** (19%) were isolated. This reaction was cleaner than that with formaldehyde, and the purification of the reaction crude was easier (*Scheme 1*).

Starting from the *anti*-isomer **2b** (*Scheme 2*), we succeeded in obtaining the fourth stereoisomer, trimer **9**. Thus, hydrogenation of **2b** over Pd/C afforded the corresponding amino derivative **13** in 69% yield. The latter was transformed into amide **14** in 62% yield by reaction with 6-nitroisatoic anhydride in THF, and amide **14** was then reduced in 45% yield to amine **15** using BH₃/Me₂S in THF. Treatment of **15** with HMTA in TFA at room temperature for 48 h yielded a mixture from which **9** (33%) and **6** (19%) were isolated.

Compound 7 was quantitatively isomerized at 80° in HCl to a 1:2:2:2 mixture of 9/6/7/8 (according to 1 H-NMR analysis). This result is similar to those obtained in isomerization experiments of the dimeric bases 2 [5] and 4 [7].

2. Structure Elucidation. The structures of the new trimeric $Tr\ddot{o}ger$'s bases were established by HR-MS, and 1 H- and 13 C-NMR (NOE, COSY, HMQC, HMBC) experiments based on previous studies of related $Tr\ddot{o}ger$'s bases [9] [10]. The cyclization of amines **12** and **15** is always regioselective, as in the precedent cases [5-7] [11], and the orientation of the cyclization has been determined by 1 H-NMR. H-C(9) and H-C(10) form an AB system, with a 3J value of 8.7 Hz, corresponding to an ortho arrangement of aromatic H-atoms in **6-8**, and as an A_2 system in **9**. However, a NOE-

Scheme 1

i) SnCl₂, HCl, EtOH, reflux, 1 h. ii) 6-Nitroisatoic anhydride²), THF, reflux, 4 h. iii) BH₃· Me₂S, THF, reflux, 8 h. iv) 37% aq. CH₂O, 95% EtOH, 36% aq. HCl, 90°, 40 h. v) Hexamethylenetetramine (HMTA), CF₃CO₂H (TFA), r.t., 72 h.

DIF experiment showed a correlation between H_x –C(18) and H_n –C(19) establishing that **9** is regiochemically identical to the other three isomers. The isomerization experiment also showed that the four bases must have the same connectivity.

We unequivocally assigned the relative configurations of **6**, **7**, **8**, and **9**, as *anti,syn*, *syn,anti*, *syn,syn* and *anti,anti*, repectively, based on the relative spatial disposition of contiguous methylene bridges, from their 1 H-NMR spectra (*Figure*). As shown in *Table 1*, long-range COSY experiments indicated a homoallylic ${}^{5}J$ coupling between the CH₂ H-atoms linked to the same *internal* aromatic ring, *i.e.*, H-C(6) and H-C(7) in **2a** and **2b**, whose structures were previously established [5]. In the *syn* arrangement of **2a**, a ${}^{5}J$ coupling is observed between the *exo* H-atom of one CH₂ H-atom and the *endo* H-atom of the other one. In the *anti*-isomer **2b**, homoallylic coupling is observed between both *exo* or *endo* protons. Likewise, we had observed a NOE interaction between H_x -C(6) and H_x -C(7) in the *syn* isomer **2a**, and between the *exo* H-atom of one CH₂ group, *e.g.*, H_x -C(6), and the *endo* H-atom of the other one, *e.g.*, H_n -C(7), in the *anti* arrangement of **2b**.

We have determined the homoallylic and, when possible, NOE correlations between the internal CH_2 resonances of the four diastereoisomeres 6-9 (*Table 1*) and,

Scheme 2

i) H2, Pd/C, EtOH, r.t., 4 h. ii) 6-Nitroisatoic anhydride²), THF, reflux, 6 h. iii) BH3 · Me2S, THF, reflux, 10 h. $i\nu)$ HMTA, TFA, r.t., 48 h.

thus, could unequivocally assign the relative configurations. This assignment was also in agreement with the observation that, at room temperature, syn-2a gives rise to 7 and 8, whereas anti-2b affords 6 and 9, because isomerization of the starting dimeric base takes place only at high temperature [5-7].

Table 1. Homoallylic Coupling and NOE Correlations Observed in Dimeric and Trimeric Tröger's Bases. At 500 MHz in (D_6) acetone.

Compound	^{5}J [Hz]	NOE	Rel. configuration
2a	6n,7x	6x/7x	syn
2 b	6x,7x	6x/7n 7x/6n	anti
6	6x,7x 18x,19n	18x/19x	anti syn
7	6x,7n 18x,19x	19n/18x	syn anti
8	6x,7n 18x,19n	18x/19x	syn syn
9	6x,7x 18x,19x	18x/19n	anti anti

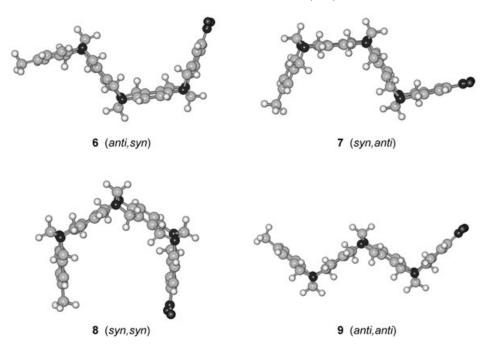


Figure. Computer-generated structures of the four possible diastereoisomers 6-9

It has been reported [5] that, in dimeric Tr"oger's bases, the aromatic H-atoms in the external rings are more shielded in syn-2a than in anti-2b in the corresponding 1 H-NMR spectra (CDCl₃). This characteristic could also be used to determine the configurations of 6-9. As evident from $Table\ 2$, these H-atoms are shielded and deshielded most in 8 and 9, respectively, and were, thus, assigned syn,syn and anti,anti configuration, respectively. For 6 and 7, we observed that H-C(1) and H-C(3) in 6 are more deshielded than in 7, but H-C(15) is more shielded, so we could assign anti,syn-and syn,anti-configurations, respectively. This criterion is less conclusive than the previous one, based on the homoallylic coupling and NOE correlations, because the chemical shifts are dependent on the solvent and on the concentration of the sample, which requires, for an unequivocal assignment, the presence of all stereoisomers, as in this work.

Table 2. ¹H-NMR Chemical Shifts (in ppm) of Aromatic, External-Ring H-Atoms. At 500 MHz in (D₆)acetone.

Compound	H-C(1)	H-C(3)	H-C(4)	H-C(13)	H-C(15)	H-C(16)
2a	6.61	6.82	6.88	7.78 ^a)	7.89 ^b)	7.27°)
2b	6.72	6.93	6.97	7.91 ^a)	8.00 ^b)	7.36 ^b)
6	6.68	6.90	6.92	7.83	7.92	7.30
7	6.65	6.86	6.93	7.83	7.95	7.30
8	6.45	6.71	6.79	7.69	7.81	7.15
9	6.72	6.92	6.98	7.90	7.99	7.35

^a) H-C(12). ^b) H-C(10). ^c) H-C(9).

Conclusions. – We have reported the successful synthesis of the four new chiral molecular tweezers 6-9. Compound 8, with its cage structure, is the first example of a heterocyclic chiral molecular clip planed to be used in host – guest chemistry in the near future.

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Experimental Part

General. 6-Nitroisatoic anhydride²) was prepared from 5-nitroisatin according to a literature procedure [12], and was used without further purification. Melting points (m.p.) are uncorrected. IR Spectra: in cm⁻¹. 1 H-NMR Spectra were recorded at 200, 300, and 500 MHz, and 13 C-NMR spectra were recorded at 50, 75, and 125 MHz; chemical shifts δ in ppm rel. to internal Me₄Si, coupling constants *J* in Hz.

General Procedure (GP 1) for the Hydrogenation of Compounds 2. The appropriate substrate 2 in EtOH was hydrogenated over 10% Pd/C at r.t. and at a pressure of 2 bar during the time indicated in each case. The mixture was filtered through Celite, which was washed with EtOH, and the resulting filtrate was evaporated to dryness under reduced pressure to afford the corresponding amino compounds after flash chromatography (FC) (SiO₂; AcOEt/MeOH 8:2).

syn-23-Methyl-1,6,14,19-tetraazaheptacyclo[17.7.1.1 6,14 .02-17.05-16.0 8,13 .02 0,25 Joctacosa-2,4,8,10,12,16,20,22,24-nonaen-10-amine (10). Prepared according to $GP\ 1$ from 2a (300 mg, 0.71 mmol) in EtOH (45 ml) and Pd/C (150 mg) for 7 h. Yield: 42%. M.p. 203 – 205° (dec.). $R_{\rm f}$ (AcOEt/MeOH 8:2) 0.16. IR (KBr): 3420, 3348, 3200, 2941, 2891, 1622, 1497, 1475, 1342, 1213, 1069, 968, 949, 831. 1 H-NMR (CDCl₃): 2.21 (s, 3 H); 3.78 (d, J = 16.8, 1 H); 3.85 (d, J = 17.1, 1 H); 4.02 (d, J = 16.8, 1 H); 4.09 (d, J = 16.6, 1 H); 4.19 – 4.21 (m, 4 H); 4.34 (d, J = 16.8, 1 H); 4.38 (d, J = 17.1, 1 H); 4.58 (d, J = 16.8, 1 H); 4.63 (d, J = 16.6, 1 H); 6.18 (d, J = 2.6, 1 H); 6.47 (dd, J = 8.3, 2.6, 1 H); 6.70 (br. s, 1 H); 6.90 (d, J = 8.3, 1 H); 6.92 – 7.02 (m, 4 H). 13 C-NMR (CDCl₃): 20.8; 55.77; 55.83; 57.8; 57.9; 66.4; 66.6; 112.3; 115.0; 124.1 (2 C); 124.8; 124.9; 125.0; 126.0; 127.1; 127.6; 128.2; 128.7; 133.6; 139.6; 142.8; 143.7; 143.8; 145.7. Anal. calc. for C_{25} H₂₅N₅: C 75.92, H 6.37, N 17.71; found: C 75.83, H 6.27, N 17.69.

anti-23-Methyl-1,6,14,19-tetraazaheptacyclo[17.7.1. 6,14 .0 2,17 .0 5,16 .0 8,13 .0 20,25]octacosa-2,4,8,10,12,16,20,22,24-nonaen-10-amine (13). Prepared according to $GP\ 1$ from **2b** (100 mg, 0.24 mmol) in EtOH (25 ml) and Pd/C (75 mg) for 4 h. Yield: 69%. M.p. 260–265° (dec.). $R_{\rm f}$ (AcOEt/MeOH 8:2) 0.26. IR (KBr): 3339, 3215, 2947, 2899, 1618, 1497, 1474, 1329, 1217, 1074, 974, 961, 841. 14 H-NMR (CDCl₃): 2.24 (s, 3 H); 3.78 (d, J = 17.1, 1 H); 3.86 (d, J = 16.9, 1 H); 4.04 (d, J = 16.8, 1 H); 4.07 –4.23 (m, 5 H); 4.27 (d, J = 17.1, 1 H); 4.32 (d, J = 16.9, 1 H); 4.56 (d, J = 16.8, 1 H); 4.62 (d, J = 16.6, 1 H); 6.22 (d, J = 2.5, 1 H); 6.53 (dd, J = 8.3, 2.5, 1 H); 6.74 (br. s, 1 H); 6.94 (d, J = 8.3, 1 H); 6.97 –7.08 (m, 4 H). 13 C-NMR (CDCl₃): 20.9; 55.6; 55.8; 58.0; 58.1; 66.2; 66.4; 112.3; 115.1; 124.2; 124.4; 124.5; 124.8; 124.9; 125.7; 127.2; 127.3; 128.3; 134.1; 143.7; 144.9. Anal. calc. for $C_{25}H_{25}N_5$: C 75.92, H 6.37, N 17.71; found: C 76.10, H 6.18, N 17.53.

Reduction of 2a with Stannous Chloride. To a mixture of 2a (80 mg, 0.19 mmol) and $SnCl_2 \cdot 2 H_2O$ (186 mg, 0.83 mmol) were added, at r.t. under Ar gas, anh. EtOH (1.5 ml) and then 36% aq. HCl (1.5 ml). The mixture was refluxed for 1 h and cooled to r.t. Then, 12m aq. NaOH soln. (15 ml) was added, the mixture was refluxed for 1 h, kept overnight at r.t., and extracted with CH_2Cl_2 (3 × 15 ml). The org. layer was washed with H_2O (20 ml), dried (MgSO₄), and evaporated under reduce pressure to yield almost pure 10 quantitatively.

General Procedure (GP 2) for the Synthesis of Amides 11 and 14. 6-Nitroisatoic anhydride²) was added in small portions at r.t. and Ar atmosphere to a soln. of the free amine (10 or 13) in anh. THF. The mixture was heated at reflux for the time indicated, and evaporated to dryness under reduced pressure. FC (SiO_2) of the crude afforded the pure amides.

2-Amino-N-(syn-23-methyl-1,6,14,19-tetraazaheptacyclo[17.7.1.1 6,14 ,0 2,17 ,0 5,16 ,0 8,13 ,0 20,25]octacosa-2,4,8,10,12, 16,20,22,24-nonaen-10-yl)-5-nitrobenzamide (**11**). Prepared according to *GP* 2 from **10** (1.15 g, 2.91 mmol) in THF (8 ml) and 6-nitroisatoic anhydride²) (605 mg, 2.91 mmol) for 4 h, followed by FC (SiO₂; AcOEt/MeOH 98:2). Yield: 66%. M.p. > 350° (dec.). IR (KBr): 3450, 3331, 2941, 2893, 1655, 1618, 1589, 1497, 1474, 1323, 1213, 1126, 1105, 833. 1 H-NMR ((D₆)DMSO): 2.08 (s, 3 H); 3.85 (2d, 2 H); 3.96 (d, 1 H); 4.00 (d, 1 H); 4.08 -4.13 (m, 4 H); 4.30 (2d, 2 H); 4.50 (d, J = 16.4, 1 H); 4.56 (d, J = 16.8, 1 H); 6.63 (br. s, 1 H); 6.80 (d, J = 9.3, 1 H); 6.87 -6.98 (m, 3 H); 7.04 (d, J = 8.6, 1 H); 7.27 (d, J = 2.2, 1 H); 7.36 (dd, J = 8.6, 2.2, 1 H); 7.53 (br. s, 2 H); 8.03 (dd, J = 9.3, 2.4, 1 H); 8.46 (d, J = 2.4, 1 H); 10.20 (s, 1 H). 13 C-NMR ((D₆)DMSO): 19.8; 55.09; 55.11; 570; 57.2;

 $65.37; 65.43; 113.5; 115.1; 118.3; 119.4; 122.9; 123.0; 123.9; 124.1; 124.2; 124.3; 125.59; 126.63; 126.9; 127.0; 127.1; 127.4; 131.9; 133.7; 134.4; 142.6; 142.8; 143.8; 145.2; 154.4; 165.0. Anal. calc. for <math>C_{41}H_{37}N_9O_3$: C 69.97, H 5.30, N 17.91; found: C 70.03, H 5.21, N 17.88.

2-Amino-N-(anti-23-methyl-1,6,14,19-tetraazaheptacyclo[17.7.1. 6,14 .02- 17 .05- 16 .08- 13 .020- 25]octacosa-2,4,8,10,12, 16,20,22,24-nonaen-10-yl)-5-nitrobenzamide (**14**). Prepared according to *GP* 2 from **13** (70 mg, 0.18 mmol) in THF (1.5 ml) and 6-nitroisatoic anhydride²) (55 mg, 0.26 mmol) for 6 h, followed by FC (SiO₂; AcOEt/MeOH 9:1). Yield: 62%. M.p. 360 – 364° (dec.). IR (KBr): 3449, 3339, 2930, 2897, 1655, 1618, 1591, 1497, 1475, 1325, 1219, 1190, 1128, 837. 14 H-NMR ((D₆)DMSO): 2.15 (*s*, 3 H); 3.78 (*d*, *J* = 16.7, 1 H); 3.81 (*d*, *J* = 16.8, 1 H); 3.95 – 4.14 (*m*, 6 H); 4.31 (2*d*, 2 H); 4.49 (*d*, *J* = 16.3, 1 H); 4.55 (*d*, *J* = 16.3, 1 H); 6.72 (br. *s*, 1 H); 6.82 (*d*, *J* = 9.3, 1 H); 6.89 – 6.98 (*m*, 4 H); 7.07 (*d*, *J* = 8.7, 1 H); 7.30 (*d*, *J* = 2.2, 1 H); 7.39 (*dd*, *J* = 8.7, 2.2, 1 H); 7.62 (br. *s*, 2 H); 8.05 (*dd*, *J* = 9.3, 2.6, 1 H); 8.54 (*d*, *J* = 2.6, 1 H); 10.29 (*s*, 1 H). 13 C-NMR ((D₆)DMSO): 19.9; 54.85; 54.86; 57.3; 57.5; 65.1; 65.2; 113.0; 115.4; 118.6; 119.8; 123.07; 123.14; 124.2; 124.4; 124.5; 124.7; 125.7; 126.5; 127.1; 127.2; 127.3; 127.6; 131.8; 133.6; 134.4; 142.8; 143.0; 144.0; 145.3; 154.7; 165.3. Anal. calc. for C₄₁H₃₇N₉O₃: C 69.97, H 5.30, N 17.91; found: C 70.19, H 5.33, N 17.69.

General Procedure (GP 3) for the Reduction of the Amides 11 and 14. A 10 $\rm M$ BH $_3$ · Me $_2$ S soln. (220 $\rm \mu l$, 2.2 mmol) was added dropwise at 0 $^\circ$ under Ar gas to a suspension of the corresponding amide (0.37 mmol) in anh. THF (3 ml). The mixture was heated at reflux for the time indicated. Then. 6 $\rm M$ aq. HCl soln. (14 ml) was added dropwise at 0 $^\circ$, and the mixture was stirred at r.t. for 2–3 h. The soln. was made basic at 0 $^\circ$ with 6 $\rm M$ aq. NH $_3$ soln. (pH 11), and extracted with CH $_2$ Cl $_2$ (3 × 15 ml). The org. layer was washed with H $_2$ O (15 ml), dried (MgSO $_4$), and evaporated under reduced pressure. The residue was subjected to FC (SiO $_2$) to afford the pure amine (12 or 15).

 $syn-N-(2-Amino-5-nitrobenzyl)-23-methyl-1,6,14,19-tetraazaheptacyclo[17.7.1.1^{6,14}.0^{2,17}.0^{5,16}.0^{8,13}.0^{20,25}] octacosa-2,4,8,10,12,16,20,22,24-nonaen-10-amine (12). Prepared according to $GP 3$, for 3 h, followed by FC (SiO_2; AcOEt/MeOH 9:1). Yield: 71%. M.p. 225 – 228° (dec.). IR (KBr): 3420, 3352, 2939, 2891, 1618, 1585, 1497, 1474, 1439, 1313, 1213, 1101, 833. <math display="inline">^1$ H-NMR (CDCl_3): 2.23 (s, 3 H); 3.79 (d, J=16.9,1 H); 3.85 (d, J=17.1,1 H); 4.06 (d, J=17.1,1 H); 4.09 (d, J=16.6,1 H); 4.11 – 4.21 (m, 4 H); 4.36 (d, J=16.9,1 H); 4.39 (d, J=17.1,1 H); 4.62 (d, J=16.6,1 H); 4.64 (d, J=17.1,1 H); 4.98 (br. s, 2 H); 6.26 (d, J=2.6,1 H); 6.54 (dd, J=8.5,2.6,1 H); 6.61 (d, J=9.5,1 H); 6.72 (br. s, 1 H); 6.92 – 7.03 (m, 5 H); 8.01 – 8.07 (m, 2 H). 13 C-NMR (CDCl_3): 20.8; 477; 55.77; 55.82; 57.8; 58.0; 66.3; 66.5; 111.2; 114.31; 114.34; 121.4; 124.1 (2 C); 124.77; 124.82; 125.1; 125.6; 126.2; 126.3; 127.1; 127.7; 128.1; 129.0; 133.6; 138.6; 140.4; 143.73; 143.76; 144.2; 145.8; 152.3. Anal. calc. for \$C_4_1H_{39}N_9O_2: C 71.39, H 5.70, N 18.27; found: C 71.20, H 5.81, N 18.11.

anti-N-(2-Amino-5-nitrobenzyl)-23-methyl-1,6,14,19-tetraazaheptacyclo[17.7.1. $^{1.04}$,0 $^{2.17}$,0 $^{5.16}$,0 $^{8.13}$,0 $^{20.25}$]octacosa-2,4,8,10,12,16,20,22,24-nonaen-10-amine (**15**). Prepared according to GP 3, for 2 h, followed by FC (SiO₂; AcOEt). Yield: 45%. M.p. 240 – 243° (dec.). IR (KBr): 3440, 3377, 2930, 2897, 1618, 1585, 1497, 1474, 1437, 1327, 1217, 1190, 1099, 841. 14 H-NMR (CDCl₃): 2.25 (s, 3 H); 3.80 (d, J = 16.6, 1 H); 3.86 (d, J = 16.7, 1 H); 4.06 – 4.20 (m, 6 H); 4.30 (d, J = 16.6, 1 H); 4.33 (d, J = 16.7, 1 H); 4.62 (d, J = 16.9, 1 H); 4.63 (d, J = 16.9, 1 H); 4.99 (br. s, 2 H); 6.30 (d, J = 2.5, 1 H); 6.58 – 6.66 (m, 2 H) 6.74 (br. s, 1 H); 6.98 – 7.07 (m, 5 H); 8.03 – 8.08 (m, 2 H). 13 C-NMR (CDCl₃): 20.9; 47.6; 55.87; 55.91; 58.3; 58.5; 66.4; 66.5; 111.3; 114.3; 114.4; 121.5; 124.27; 124.29; 124.9; 125.02; 125.05; 125.6; 126.1; 126.3; 127.3; 127.6; 128.2; 129.0; 133.8; 138.8; 140.4; 143.76; 143.83; 144.2; 145.6; 152.2. Anal. calc. for C₄₁H₃₉N₉O₂: C 71.39, H 5.70, N 18.27; found: C 71.47, H 5.92, N 18.33.

Synthesis of the Trimeric Diastereoisomers 6-8 with Formaldehyde. At r.t. under Ar gas, to a stirred suspension of 12 (180 mg, 0.33 mmol) in 95% EtOH (1 ml) were successively added 35-40% aq. formaldehyde soln. (147 μ l) and 36% aq. HCl soln. (169 μ l). The mixture was heated at reflux for 40 h, cooled to r.t., and made basic with conc. aq. NH₃ soln. (pH 11). The alkaline soln. was extracted with CH₂Cl₂ (3×10 ml), and the combined org. extracts were washed with H₂O (15 ml), dried (MgSO₄), and evaporated under reduced pressure. The crude was purified by FC (SiO₂; AcOEt/MeOH 98:2) to yield, in this order, 6 (11%), 7 (7%), and 8 (8%).

Data of anti,syn-10-Methyl-28-nitro-1,6,14,19,24,32-hexaazadecacyclo[17.17.1. $^{6.14}$, 124,32 , $^{02.17}$, $^{05.10}$, 08,13 , 020,33 , 023,34 , 026,31]nonatriaconta-2,4,8,10,12,16,20,22,26,28,30,34-dodecaene (**6**). M.p. 280 – 285° (dec.). R_1 (SiO₂; AcOEt/MeOH 8:2) 0.40. IR (KBr): 1612, 1578, 1516, 1495, 1472, 1340, 1219, 1204, 1074, 961, 939, 841. 14 H-NMR ((D₆)acetone): 2.15 (s, 3 H); 3.80 (d, J = 16.6, 1 H); 3.82 (d, J = 16.8, 1 H); 3.92 (d, J = 11.4, 1 H); 3.99 (d, J = 17.1, 2 H); 4.01 (s, 2 H); 4.02 (d, J = 11.4, 1 H); 4.20 (d, J = 16.9, 1 H); 4.21 (d, J = 16.9, 1 H); 4.24 (s, 2 H); 4.27 (d, J = 17.1, 1 H); 4.28 (d, J = 17.0, 1 H); 4.31 (d, J = 16.6, 1 H); 4.47 (d, J = 17.1, 1 H); 4.51 (d, J = 17.2, 1 H); 4.72 (d, J = 17.0, 1 H); 6.81 (s, 2 H); 6.90 (dd, J = 8.2, 1 H); 6.92 (d, J = 8.2, 1 H); 6.98 (d, J = 8.7, 1 H); 7.30 (d, J = 8.9, 1 H); 7.83 (d, J = 2.4, 1 H); 7.92 (dd, J = 8.9, 2.5, 1 H). 13 C-NMR ((D₆)acetone): 20.8; 55.8; 56.1; 56.4; 56.8; 58.5; 59.0; 66.5; 66.76; 66.83; 122.9; 123.6; 124.68; 124.70; 125.0;

125.3; 125.5; 125.7; 126.06; 126.10; 126.2; 126.8; 127.8; 128.5; 128.8; 130.6; 133.6; 143.8; 144.2; 144.82; 144.86; 145.4; 147.1; 156.5. HR-FAB-MS: 569.2549 ($C_{34}H_{31}N_7O_2$): 570.2616 ($[M+1]^+$, $C_{34}H_{32}N_7O_2^+$; calc. 570.2618).

Data of **7**. M.p. $265-267^{\circ}$ (dec.). R_1 (AcOEt/MeOH 8:2) 0.35. IR (KBr): $1611, 1582, 1516, 1495, 1472, 1340, 1221, 1202, 1072, 962, 939, 841. <math>^{1}$ H-NMR ((D₆)acetone): 2.09 (s, 3 H); 3.81 (d, J = 16.8, 1 H); 3.94 (d, J = 17.7, 1 H); 3.96 (d, J = 17.1, 1 H); 3.97 (d, J = 12.4, 1 H); 4.02 (d, J = 16.4, 1 H); 4.03 (d, J = 11.7, 1 H); 4.05 (d, J = 16.4, 1 H); 4.09 (s, 2 H); 4.12 (d, J = 12.4, 1 H); 4.16 (d, J = 12.4, 1 H); 4.21 (d, J = 17.0, 1 H); 4.30 (d, J = 17.2, 2 H); 4.34 (d, J = 17.1, 2 H); 4.55 (d, J = 16.7, 1 H); 4.63 (d, J = 17.0, 1 H); 6.65 (br. s, 1 H); 6.85 (d, J = 8.6, 1 H); 6.86 (pseudo-d, J = 6.7, 1 H); 6.88 (d, J = 8.7, 1 H); 6.93 (pseudo-d, J = 9.8, 2 H); 6.96 (d, J = 8.7, 1 H); 7.30 (d, J = 8.9, 1 H); 7.83 (d, J = 2.5, 1 H); 7.95 (dd, J = 8.9, 2.6, 1 H). 13 C-NMR ((D₆)acetone): 20.7; 55.9; 56.2; 56.5; 56.7; 58.7; 58.9; 66.2; 66.5; 67.3; 122.9; 123.6; 124.6; 124.8; 124.9; 125.4; 125.6; 125.8; 125.97; 126.01; 126.3; 126.6; 127.8; 128.6; 128.9; 130.5; 133.7; 143.9; 144.2; 144.8; 144.9; 145.4; 147.2; 156.6. HR-FAB-MS: 569.2531 (C₃₄H₃₁N₇O₂): 570.2594 ([M + 1]+, C₃₄H₃₂N₇O₂+; calc. 570.2618).

Data of **8**. M.p. 250 – 252°. $R_{\rm f}$ (SiO₂; AcOEt/MeOH 8:2) 0.20. IR (KBr): 1612, 1582, 1512, 1497, 1472, 1339, 1219, 1070, 970, 947, 835. $^{\rm i}$ H-NMR ((D₆)acetone): 1.98 (s, 3 H); 3.88 (d, J = 16.9, 1 H); 3.89 (d, J = 16.7, 1 H); 3.95 (d, J 17.1, 1 H); 4.01 (pseudo-d, J = 16.0, 1 H); 4.03 (s, 2 H); 4.04 (pseudo-d, J = 14.01, 1 H); 4.08 (d, J = 11.2, 1 H); 4.11 (d, J = 16.9, 1 H); 4.14 (pseudo-d, J = 14.7, 1 H); 4.16 (s, 2 H); 4.29 (d, J = 17.1, 1 H); 4.31 (d, J = 16.9, 1 H); 4.32 (d, J = 17.2, 1 H); 4.46 (pseudo-d, J = 15.8, 2 H); 4.63 (d, J = 16.9, 1 H); 6.45 (br. s, 1 H); 6.71 (br. d, J = 8.1, 1 H); 6.79 (d, J = 8.1, 1 H); 6.85 (s, 2 H); 6.89 (d, J = 8.6, 1 H); 6.91(d, J = 8.8, 1 H); 7.15 (d, J = 8.8, 1 H); 7.69 (d, J = 2.2, 1 H); 7.81 (dd, J = 8.8, 2.4, 1 H). 13 C-NMR ((D₆)acetone): 20.6; 55.7; 55.8; 56.4; 56.5; 58.2; 58.6; 66.6; 66.7; 67.2; 122.7; 123.6; 124.4; 124.6; 125.0; 125.3; 125.4; 125.7; 125.77; 125.79; 126.1; 126.6; 127.8; 128.38; 128.39; 130.0; 133.8; 143.8; 144.2; 144.72; 144.74; 145.4; 146.8; 156.1. HR-FAB-MS: 569.2540 (C₃₄H₃₁N₇O₂): 570.2610 ([M + 1] $^+$, C₃₄H₃₂N₇O $_2^+$; calc. 570.2618).

Synthesis of the Trimeric Diastereoisomers 6, 8, and 9 with HMTA in TFA. Hexamethylenetetramine (HMTA; 1.92 mmol) was added at 0° under Ar gas to a suspension of the corresponding amine (12 or 15; 1.75 mmol) in anh. trifluoroacetic acid (TFA; 15 ml). The mixture was stirred at r.t. during the time indicated, and poured on cold H_2O (25 ml). The soln. was carefully made alkaline at 0° with 25% aq. NH₃ soln. (pH 11), and extracted with CH_2Cl_2 (3 × 50 ml). The org. layer was washed with H_2O (50 ml), dried (MgSO₄), and evaporated under reduced pressure. The crude was purified by FC (SiO₂). From 12, after 72 h, followed by FC (AcOEt/MeOH 93:7), isomers 7 (46%) and 8 (19%) were isolated. From 15, after 48 h, followed by FC ($CH_2Cl_2/MeOH$ 98.5:1.5), 9 (33%) and 6 (19%) were isolated in this order, resp.

 $Data\ of\ \textbf{9}.\ \text{M.p.}\ 315-318^\circ\ (\text{dec.}).\ R_f\ (\text{AcOEt/MeOH}\ 8:2)\ 0.35.\ IR\ (\text{KBr}):\ 1611,\ 1580,\ 1516,\ 1495,\ 1472,\ 1339,\ 1221,\ 1202,\ 1074,\ 961,\ 939,\ 845.\ ^1\text{H-NMR}\ ((D_6)\text{acetone}):\ 2.17\ (s,\ 3\ H);\ 3.88\ (\text{pseudo-}d,\ J=16.6,\ 3\ H);\ 3.97\ (d,\ J=12.4,\ 1\ H);\ 4.01\ (d,\ J=12.7,\ 1\ H);\ 4.07\ (d,\ J=16.7,\ 1\ H);\ 4.11\ (\text{pseudo-}d,\ J=15.2,\ 2\ H);\ 4.16\ (d\ J=12.4,\ 1\ H);\ 4.22\ (s,\ 2\ H);\ 4.28\ (\text{pseudo-}d,\ J=19.1,\ 1\ H);\ 4.30\ (\text{pseudo-}d,\ J=18.6,\ 1\ H);\ 4.32\ (d,\ J=17.0,\ 1\ H);\ 4.34\ (d,\ J=17.5,\ 1\ H);\ 4.51\ (d,\ J=16.9,\ 1\ H);\ 4.51\ (d,\ J=16.9,\ 1\ H);\ 6.72\ (\text{br.}\ s,\ 1\ H);\ 6.92\ (d,\ J=8.6,\ 1\ H);\ 6.93\ (\text{pseudo-}d,\ J=5.6,\ 1\ H);\ 6.95\ (d,\ J=9.5,\ 1\ H);\ 6.96\ (s,\ 2\ H);\ 6.98\ (d,\ J=8.1,\ 1\ H);\ 7.35\ (d,\ J=8.9,\ 1\ H);\ 7.90\ (d,\ J=2.5,\ 1\ H);\ 7.99\ (dd,\ J=8.9,\ 2.5,\ 1\ H).\ ^{13}\text{C-NMR}\ ((D_6)\text{acetone}):\ 20.8;\ 56.1;\ 56.2;\ 56.5;\ 56.7;\ 58.7;\ 59.0;\ 66.4;\ 66.6;\ 67.11;\ 123.0;\ 123.6;\ 124.82;\ 124.85;\ 124.9;\ 125.4;\ 125.68;\ 125.74;\ 126.1;\ 126.2;\ 126.4;\ 126.6;\ 127.9;\ 128.4;\ 128.6;\ 130.6;\ 133.7;\ 143.9;\ 144.2;\ 144.8;\ 144.9;\ 145.3;\ 147.1;\ 156.6.\ HR-FAB-MS:\ 569.2551\ (C_{34}H_{31}N_7O_2):\ 570.2623\ ([M+1]^+,\ C_{34}H_{32}N_7O_2^+;\ \text{calc.}\ 570.2618).$

Isomerization of 7. A soln. of 7 (20 mg, 0.035 mmol) in 95% EtOH (2 ml) containing 36% aq. HCl soln. (150 μ l, 1.8 mmol) was stirred at 80° for 36 h. The mixture was cooled to r.t., poured on H₂O, made basic with 25% aq. NH₃ soln. (pH 11), and extracted with CH₂Cl₂. The org. layer was washed with H₂O, dried (MgSO₄), and evaporated under reduced pressure to afford, in quantitative yield, a 9/6/7/8 1:2:2:2 mixture of diastereoisomers (determined by 1 H-NMR).

REFERENCES

- [1] J. Tröger, J. Prakt. Chem. 1887, 36, 225.
- [2] 'Fascinating Molecules in Organic Chemistry', Ed. F. Vögtle, John Wiley, Chichester, 1992, p. 237.
- [3] a) S. Paliwal, S. Geib, C. S. Wilcox, J. Am. Chem. Soc. 1998, 120, 11192; b) E.-I. Kim, S. Paliwal, C. S. Wilcox, J. Am. Chem. Soc. 1994, 116, 4497; c) M. Demeunynck, A. Tatibouët, 'Recent Developments in Tröger's Base Chemistry', in 'Progress in Heterocyclic Chemistry', Eds. G. W. Gribble, T. L. Gilchrist, Pergamon Press, Oxford, 1999, Vol. 11, p. 1; d) M. Harmata, M. Kahraman, Tetrahedron: Asymmetry 2000, 11, 2875; e) J. Jensen, M. Strozyc, K. Wärnmark, Synthesis 2002, 2761; f) J. Jensen, J. Tejler, K. Wärnmark,

- J. Org. Chem. 2002, 67, 6008; g) A. Hannson, J. Jensen, O. F. Wendt, K. Wärnmark, Eur. J. Org. Chem. 2003, 3179; h) M. Miyake, C. S. Wilcox, Heterocycles 2002, 56, 515; i) F. Hof, D. M. Scofield, W. B. Schweizer, F. Diederich, Angew. Chem., Int. Ed. 2004, 43, 5056; j) S. Sergeyev, F. Diederich, Angew. Chem., Int. Ed. 2004, 43, 1738.
- [4] M. Harmata, C. L. Barnes, Tetrahedron Lett. 1990, 31, 1825; M. Harmata, C. L. Barnes, J. Am. Chem. Soc. 1990, 112, 5655; J. Fleischhauer, M. Harmata, M. Kahraman, A. Koslowski, C. J. Welch, Tetrahedron Lett. 1997, 38, 8655.
- [5] C. Pardo, E. Sesmillo, E. Gutiérrez-Puebla, A. Monge, J. Elguero, A. Fruchier, J. Org. Chem. 2001, 66, 1607.
- [6] T. Mas, C. Pardo, F. Salort, J. Elguero, M. R. Torres, Eur. J. Org. Chem. 2004, 1097.
- [7] T. Mas, C. Pardo, J. Elguero, Arkivoc 2004, iv, 86.
- [8] F. G. Klärner, M. Lobert, U. Naatz, H. Bandmann, R. Boese, Chem.–Eur. J. 2003, 9, 5036; F. G. Klärner, B. Kahlert, Acc. Chem. Res. 2003, 36, 919.
- [9] J. Cudero, P. Jiménez, C. Marcos, C. Pardo, M. Ramos, J. Elguero, A. Fruchier, Magn. Reson. Chem. 1996, 34, 318.
- [10] C. Pardo, M. Ramos, A. Fruchier, J. Elguero, Magn. Reson. Chem. 1996, 34, 708.
- [11] M. Valik, B. Dolensky, H. Petricková, V. Král, Collect. Czech. Chem. Commun. 2002, 67, 609.
- [12] G. Reissenweber, D. Mangold, Angew. Chem., Int. Ed. 1980, 19, 222.

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